A CONVENIENT PREPARATION OF C-TERMINAL PEPTIDE ALCOHOLS BY SOLID PHASE SYNTHESIS

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Summary: A new method is described which allows ready access to C-terminal pepride alcohols via solid phase synthesis. The procedure involves coupling of a hemisuccinate derived from a Nprotected β -aminoalcohol to a solid phase resin, elaboration of the desired peptide by conventional synthesis and cleavage of the resulting succinate with either ammonia or hydrazine.

Peptide alcohols occur in nature and are also of interest as analogs of bioactive peptides. These materials have generally been synthesized by hydride reduction of a peptide containing a C-terminal ester function.¹ or by a condensation strategy involving solution phase coupling of a preformed peptide with an amino alcohol.2 There are two reported routes using solid phase methodology. They involve either ammonolysis of the resin bound peptide using the appropriate β -aminoalcohol,³ or lithium aluminum hydride reduction of the resin bound peptide ester at the completion of a conventional synthesis.⁴ The limitations of these methods prompted us to consider an alternative approach in which an N-protected ß-aminoalcohol could be coupled to a solid support by esterification to a difunctional linking group, followed by conventional peptide synthesis and finally, ester cleavage to afford the target peptide alcohols.

We have found that the N-protected 6-aminoalcohols 1 react readily with succinic anhydride in the presence of 4dimethylaminopyridine (DMAP) and pyridine in DMF to give the corresponding hemisuccinates 2 (Table 1). These monoacids were coupled to a benzhydrylamine resin (BHA) using diisopropylcarbodiimide DIC/HOBT, to provide the resin bound intermediates 3, which can be further elaborated by conventional solid phase peptidc synthesis. In order to illustrate this process, we have employed DIC/HOBT coupling of Boc protected aminoacids with TFA deprotections to construct a series of tetrapeptide derivatives. Liberation from the resin was accomplished by hydrolysis of the resulting succinic esters, either by treatment with ammonia/MeOH in a pressure bottle (72-96 h), or with an excess of hydrazine in DMF (24 h) to afford the tetrapeptide alcohols 4a through 4g. As indicated in Table 2, both hydrolysis methodologies provide comparable yields. The benzyl protecting group was removed from the tetrapeptide 4e by simple Pd/C catalyzed hydrogenolysis in glacial acetic acid.

The method is compatible with either Boc/OFm and Fmoc/OtBu strategies, provided that the timing of side chain deprotection is appropriate, and provides an exceptionally facile route to peptide alcohols. The application of this technology to the construction of biologically active peptide alcohols will be reported in due course.

Table **15 O-Aminoakhol-hemisuccinnates**

a. rotations were run at 25° C in ethanol (c = 1) except for 2d (CHCl₃, c = 0.55).

Table 2 Yield of Tetrapeptide Alcohols

General Procedure for the synthesis of 2:

At room temperature, compound 1 (10 mmol), succinic anhydride (3.0 g, 30 mmol, 3 eq.), DMAP (3.7 g, 30 mmol), and pyridine (3.8 mL) were stirred in 25 mL DMF (12 h). After removal of the majority of DMF under reduced pressure, the residue was dissolved in water (100 mL) and the pH brought to 9.0 using conc. NH₄OH. The mixture was washed with EtOAc (2 x 20 mL). The aqueous layer was carefully acidified to pH 3.0 using cont. HCl and then extracted with EtOAc (3 x 30 mL). Combination of the organic extracts was followed by washing with water (3 x 30 mL) and brine (30 mL). The organic layer was dried (MgSO₄), filtered, and concentrated to constant weight (0.01 torr). The crude products obtained in this way are suitable for use in the next step.

General Procedure for the synthesis of 4:

The succinates 2 (2.05 mmol) were coupled to the BHA resin⁶ (2.0 g, 0.82 mmol) using DIC (310 µL, 2.05 mmol) and HOBT (390 mg, 2.05 mmol)) in DMF (40 mL). The mixture was agitated until a negative ninhydrin test was obtained. All subsequent, deprotections, washings and couplings proceeded in the conventional manner.⁷

Ammonolysis

Ammonia (40 mL) was condensed into a chilled (-40 $^{\circ}$ C) pressure bottle containing the peptide-resin (1.5 g) and methanol (60 mL). The vessel was sealed and allowed to stir (72-96 h). After chilling the pressure bottle (-40 °C), the vessel was opened and the NH3 allowed to evaporate. The methanolic solution was then filtered and the remaining resin was washed with methanol. The filtrate was concentrated in vacua and the resulting crude residue dissolved in acetic acid and lyophilized. The crude product was redissolved in AcOH/H₂O (6 mL, 5:1) and purified by reverse phase HPLC on a Whatman C_{18} Mag 40 (40 mm x 50 cm) column using gradient elution 5% CH3CN/O.O25% TBA to 30% CH3CN/O.O25% TFA. The desired fractions were pooled and lyophilixed. Yields for the various peptide alcohols are given in Table 2.

Hydrazinolysis

The peptide resin (1.5 g) was treated with hydrazine (4.0 mL) in DMF (25 mL) with shaking (24 h). The mixture was filtered and the remaining resin was washed with DMF $(2 \times 25 \text{ mL})$. The combined filtrates were then concentrated under high vacuum (0.1 torr, bath temp 45 $^{\circ}$ C) to constant weight and sufficient Et₂O was added to the oily mass to precipitate crude solid product. The product was dissolved in AcOH/H₂O (6 mL, 5:1) and purified by mverse phase HBLC as before. Yields for the various peptide alcohols are given in Table 2.

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- 5. All compounds have been characterized by combustion analyses, IR, ¹H NMR, and mass spectra. With the exception of 2g, every compound possesses an elemental analysis within the error limits of 0.4%. Anal. Calcd for $C_{14}H_{23}N_1O_6$ (2g): C, 55.80; H, 7.69; N, 4.65. Found: C, 54.18; H, 7.58; N, 4.22; H20 content by Karl Fisher analysis, 0.39%.
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- 8. Yields based on 3.
- 9. All final peptide products were characterized by amino acid and mass spectral analysis (FAB).

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